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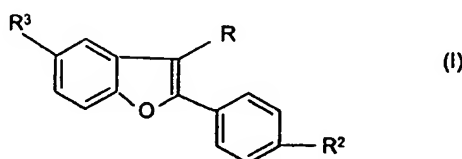
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(54) Title: COMPOUNDS FOR IMAGING ALZHEIMER'S DISEASE



(57) Abstract: The invention provides compounds of formula (I) or a salt thereof, radiolabelled versions thereof, and their use in the diagnosis or *in vivo* imaging of Amyloid-associated diseases such as Alzheimer's disease.

COMPOUNDS FOR IMAGING ALZHEIMER'S DISEASE

The present invention relates to the field of diagnostic imaging of Alzheimer's disease and provides compounds useful in such diagnostic imaging, as well as
5 methods for their preparation and use.

Alzheimer's disease is the fourth most common cause of death in the western world, after heart disease, cancer and strokes. In the USA there are approximately 4 million people suffering with Alzheimer's disease, at an annual cost of \$100 billion. Therefore, the cost per person in the USA is \$25,000 per year. There are
10 currently 20 million sufferers of dementia in the world. This is set to double to 40 million by the year 2025 as the number of people aged 65 doubles from 390 million now to 800 million in 2025. Of these 40 million, approximately 56 percent will be suffering from Alzheimer's disease, accounting for 22.2 million.

The *in vivo* imaging techniques used at present do not in all cases differentiate the
15 diagnosis of Alzheimer's disease from other forms of dementia. The differential diagnosis of patients will become increasingly important as more treatments become available. Imaging agents will also be required to image Alzheimer patients at earlier stages of the disease to allow preventive treatment, and for monitoring disease progression

20 Currently the only definitive test for Alzheimer's disease is examination of the brain at autopsy for the presence of distinctive pathophysiologies. One of the most widely acknowledged of these pathophysiologies is the presence of senile plaques in brain tissue. Senile plaques are deposits of a 40-43 amino acid protein called the β -amyloid protein. They are an early and invariant aspect of the disease and
25 it is thought that the deposition of β -amyloid occurs some time prior to the onset of clinical symptoms.

Amyloid-specific radiotracers have been suggested as potential imaging agents for Alzheimer's disease. Congo Red has been demonstrated to be an effective binder of β -amyloid, but does not cross the blood-brain barrier (BBB) well (Klunk

et al 1994 Neurobiology of Aging Vol. 15 pp. 691-698). There is no convincing functional evidence that abnormalities in the BBB reliably exist in Alzheimer's (Kalaria 1992, Cerebrovascular and Brain Metabolism Reviews, Vol 4, p 226).

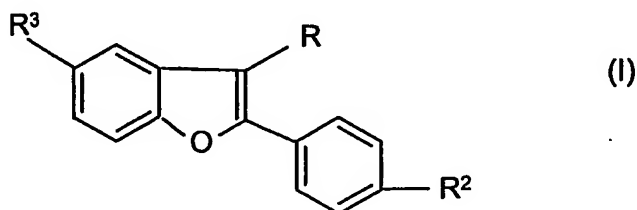
Therefore, an important property of an Alzheimer's *in vivo* imaging agent is that
5 it crosses the BBB.

US 3,947,470 and US 4,024,273 describe certain benzofuran compounds having coronary vasodilator activity. Howlett *et al* (Biochem J. (1999), 340, 283-9) describes a series of benzofuran derivatives which are inhibitors of fibril formation in the β -amyloid peptide.

10 The aim of the present invention is the provision of novel ^{99m}Tc -labelled agents for *in vivo* imaging of Alzheimer's disease. To be able to successfully image Alzheimer's disease *in vivo*, an agent must be capable of crossing the BBB as well as binding to β -amyloid.

Labelling of a compound with ^{99m}Tc requires that the compound be coupled with
15 a suitable ^{99m}Tc chelator. It is known in the art that suitable ^{99m}Tc chelators are relatively bulky molecules, and as such would act to increase the molecular weight of an amyloid-binding agent, in turn decreasing the chances of such an agent crossing the BBB. Similarly, attachment of a suitable ^{99m}Tc chelator to an amyloid-binding agent would be expected to interfere with the binding properties of the
20 agent. It is therefore surprising that we have found a class of compounds which include chelated ^{99m}Tc but which may be capable of crossing the BBB and binding with β -amyloid.

In a first aspect, this invention provides a compound of formula (I):

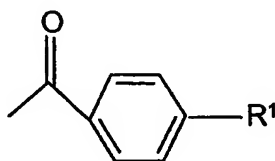


25 or a salt thereof, wherein either:

- (i) R is hydrogen and R^2 is the group $-\text{OCR}^4\text{R}^5[\text{B}]-[\text{A}]$ wherein R^4 and R^5 are independently selected from hydrogen and C_{1-6} alkyl, [A] is a chelate, and [B] is a linking group and is preferably $-\text{C}(\text{O})\text{NR}^6-$ wherein R^6 is hydrogen or C_{1-6} alkyl;

5 or

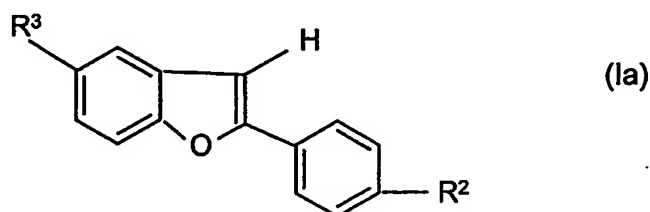
- (ii) R is the group



and one of R^1 and R^2 is the group $-\text{OCR}^4\text{R}^5[\text{B}]-[\text{A}]$ as defined above; and the other is the group $-(\text{O})_n-\text{C}_{1-6}\text{alkyl}-\text{NR}^7\text{R}^8$ wherein n is 0 or 1, and R^7 and R^8 are independently selected from hydrogen and C_{1-6} alkyl; and

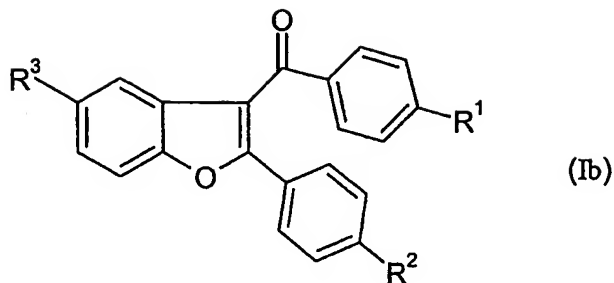
10 R^3 is halo, preferably chloro.

In one aspect of the invention, the group R in the compound of formula (I) is hydrogen, therefore there is provided a compound of formula (Ia):



or a salt thereof, wherein R^2 is the group $-\text{OCR}^4\text{R}^5[\text{B}]-[\text{A}]$ wherein R^4 and R^5 are independently selected from hydrogen and C_{1-6} alkyl, [A] is a chelate, and [B] is a linking group and is preferably $-\text{C}(\text{O})\text{NR}^6-$ wherein R^6 is hydrogen or C_{1-6} alkyl; and R^3 is halo, preferably chloro.

In a further aspect of the invention, there is provided a compound of formula (Ib):



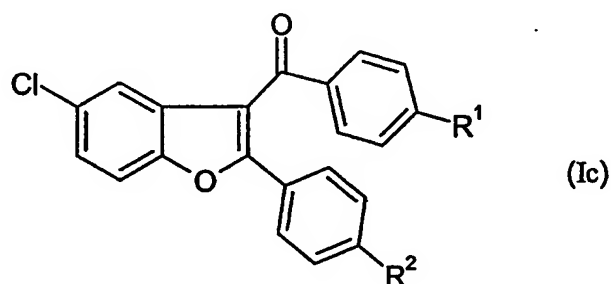
or a salt thereof, wherein:

one of R^1 and R^2 is the group $-\text{OCR}^4\text{R}^5[\text{B}]-[\text{A}]$ as for formula (I); and

the other is the group $-(\text{O})_n-\text{C}_{1-6}\text{alkyl}-\text{NR}^7\text{R}^8$ as defined for formula (I); and

5 R^3 is halo, preferably chloro.

Preferred compounds of formula (I) include those of formula (Ic):



or a salt thereof, wherein:

one of R^1 and R^2 is $-\text{OCH}_2\text{C}(\text{O})\text{NH}-[\text{A}]$ wherein $[\text{A}]$ is a chelate;

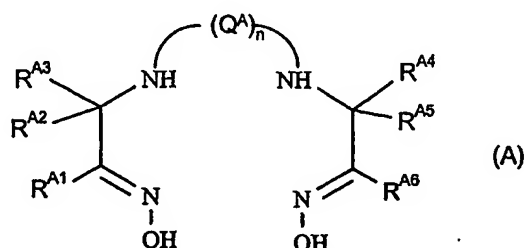
10 the other is $-\text{O}-(\text{CH}_2)_{2,3}-\text{NR}^7\text{R}^8$ wherein R^7 and R^8 are independently selected from hydrogen and C_{1-8} alkyl.

Compounds of formula (I) which are of particular interest, include those described in Examples 1(vii), 2(vii), and 8(iii), especially when Pn216 conjugated.

15

The term "chelate" is defined in the present invention as being a tetradentate metal complexing agent, preferably suitable for complexing technetium. Technetium chelates are discussed in section B of "Technetium: chemistry and radiopharmaceutical applications", Klaus Schwachau, 2000, pp.373-423, published by Wiley-Vch Verlag GmbH. Suitable chelates of the invention may be selected from:

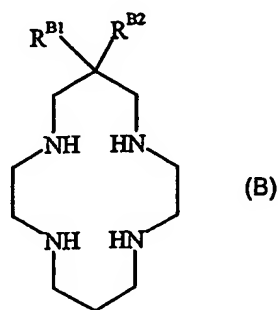
(i) diaminedioximes of formula (A):



where R^{A1} - R^{A6} are each independently selected from hydrogen, C_{1-6} alkyl, arylalkyl, alkoxyalkyl, hydroxyalkyl, haloalkyl or aminoalkyl;

and $-(Q^A)_n-$ is a bridging group wherein n is 3, 4 or 5 and each Q^A is independently selected from $-O-$, $-NR-$ and $-CR_2-$ where R is hydrogen, C_{1-6} alkyl, arylalkyl, alkoxyalkyl, hydroxyalkyl, fluoroalkyl or aminoalkyl provided that there is a maximum of one Q^A group which is selected from $-O-$ or $-NR-$.

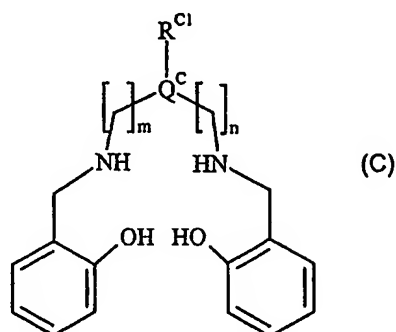
(ii) Macrocyclic N_4 ligands of formula (B):



where R^{B1} and R^{B2} are independently selected from hydrogen, C_{1-6} alkyl,

arylalkyl optionally substituted by aminoalkyl, alkoxyalkyl, hydroxyalkyl, haloalkyl or aminoalkyl.

(iii) Diaminediphenols of formula (C):

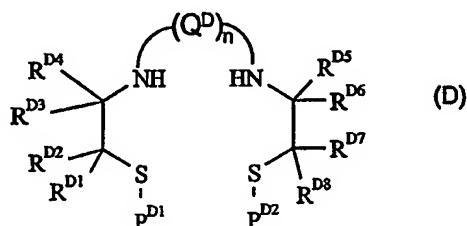


5 where Q^C is either $C(R^{C2})$ or N;

R^{C1} and R^{C2} are independently selected from hydrogen, C_{1-10} alkyl, arylalkyl, alkoxyalkyl, hydroxyalkyl, haloalkyl or aminoalkyl, and;

m is equal to n and both are integers suitably selected from 1, 2, 3, and 4.

(iv) N_2S_2 ligands of formula (D):



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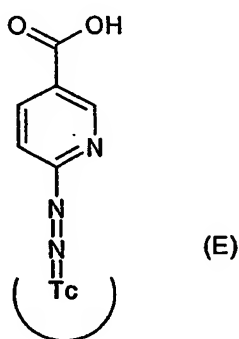
where P^{D1} and P^{D2} are hydrogen or protecting groups such as benzoyl, acetyl or ethoxyethyl which may be cleaved before or during the labelling process, and;

R^{D1} to R^{D8} are independently selected from hydrogen, C_{1-10} alkyl, arylalkyl, alkoxyalkyl, hydroxyalkyl, haloalkyl or aminoalkyl, and;

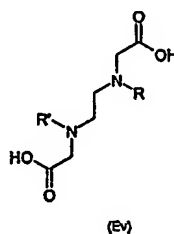
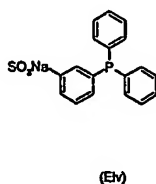
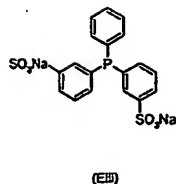
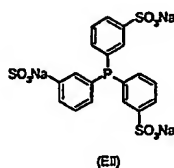
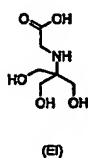
15 one or more of the pairs R^{D1}/R^{D2} , R^{D3}/R^{D4} , R^{D5}/R^{D6} , R^{D7}/R^{D8} may, together with the carbon to which they are attached, represent a $C=O$ group,

and $-(Q^D)_n-$ is a bridging group wherein n is 2, 3, 4, or 5 and each Q^D is independently selected from $-O-$, $-NR-$, and $-CR_2-$ wherein R is hydrogen, C_{1-6} alkyl, arylalkyl, alkoxyalkyl, hydroxyalkyl, haloalkyl, or aminoalkyl, provided that there is a maximum of one Q^D group which is selected from $-O-$ or $-NR-$.

- 5 (v) Hydrazino nicotinamide ligands of formula (E):



wherein technetium is co-ordinated in conjunction with a co-ligand selected from formulae (Ei) to (Ev):



- 10 wherein R^{E1} is selected from hydrogen, benzyl, and C_{1-6} alkyl (suitably methyl) and R^{E2} is selected from hydrogen and $-\text{CH}_2\text{COOH}$.

Preferred diaminedioximes of formula (A) have R^{A1} to R^{A6} independently selected from C_{1-3} alkyl, alkylaryl alkoxyalkyl, hydroxyalkyl, fluoroalkyl or aminoalkyl. In the most preferred diaminedioximes, R^{A1} to R^{A6} are all CH_3 . One preferred chelating agent of the present invention is a diaminedioxime of formula (A) where R^{A1} to R^{A6} are all CH_3 and $(Q^A)_n$ is $-\text{NH}(\text{CH}_2)_2\text{N}(\text{CH}_2\text{CH}_2\text{NH}_2)(\text{CH}_2)_2\text{NH}-$. Another preferred

chelating agent of the present invention is a diaminedioxime of formula (A) where R^{A1} to R^{A6} are all CH_3 and $(Q^A)_n$ is $-CH_2C(CH_3)(CH_2NH_2)CH_2-$. Amyloid binding agents of the invention may be suitably linked to the chelate at any functional residue. Amyloid binding agents of the invention are preferably linked via a
 5 functionality in $(Q^A)_n$, and most preferably via an amine group in $(Q^A)_n$ to form the [B] linker group $-C(O)NR^6-$.

Preferred macrocyclic N_4 ligands of formula (B) have R^{B1} as H and R^{B2} as aminoalkyl arylalkyl (suitably aminomethyl benzyl, most suitably 4-aminomethyl benzyl). Amyloid binding agents of the invention may be attached to the chelate
 10 via any functional group, but preferably via a functional group of R^{B1} or R^{B2} . Amyloid binding agents of the invention are most preferably attached to the chelate via a terminal NH_2 of R^{B2} to form the [B] linker group $-C(O)NR^6-$.

Preferred diaminediphenols of formula (C) have Q^C as $C(R^{C2})$ and one of R^{C1} and R^{C2} aminoalkyl and the other hydrogen or C_{1-6} alkyl. In a most preferred
 15 embodiment, Q^C is $C(R^{C2})$, R^{C1} is CH_3 and R^{C2} is CH_2NH_2 . Amyloid binding agents of the invention are suitably attached to any functional residue of the chelate, but preferably to either R^{C1} or R^{C2} . Amyloid binding agents of the invention are most preferably attached via a terminal NH_2 group of R^{C2} to form the [B] linker group $-C(O)NR^6-$.

20 In preferred N_2S_2 ligands of formula (D), R^{D1} , R^{D2} and R^{D7} are hydrogen and the pairs R^{D3}/R^{D4} and R^{D5}/R^{D6} , together with the carbons to which they are attached are $C=O$, and R^{D8} , is aminoalkyl. In a most preferred N_2S_2 ligand of the invention, R^{D1} , R^{D2} , R^{D7} , and R^{D8} are methyl, R^{D3} , R^{D4} , R^{D5} , R^{D6} are hydrogen, and $(Q^D)_n$ is $-CH_2CH(CH_2NH_2)-$ or $-CH_2C(CH_3)(CH_2NH_2)CH_2-$. Amyloid binding agents
 25 of the invention may be suitably attached to any functional group on the chelate, but preferably to a functional group at Q^D , and most preferably to a terminal NH_2 at Q^D to form the [B] linker group $-C(O)NR^6-$.

Synthesis of these chelate compounds is well known in the art. Diaminedioxime ligands of formula (A) are prepared by the methods described by Jurisson et al
 30 (Inorg. Chem. 1986, Vol. 25, pp. 543-549). Macrocyclic N_4 ligands of formula (B)

are prepared by the methods of (Int. J. Nuc. Med. Biol., 1984, Vol. 2(2), pp. 113-119). Diaminediphenol ligands of formula (C) may be synthesised according to the methods described by Pillai et al (Nucl. Med. Biol. 1993, Vol. 20(2), pp. 211-216). N₂S₂ ligands of formula (D) are synthesised by the methods of Sun *et al* (J. Med. Chem. 1996, Vol. 39, pp. 458-470). Hydrazino nicotinamide ligands of formula (E) are synthesised according to the methods described in US 5206370.

A "salt" of a compound of formula (I) is defined as ionic versions of the compounds of formula (I) formed by replacing one or more of the hydrogen ions of an acid with another positive ion such as an alkali metal (suitably Na⁺ or K⁺).

10 "Alkyl" used either alone or as part of another group (such as arylalkyl, hydroxyalkyl) is defined herein as any straight or branched C_nH_{2n+1} group, wherein unless otherwise specified n is 1 to 10 and preferably n is 1 to 6.

The term "halo" means a group selected from fluoro, chloro, bromo and iodo.

As stated above, for *in vivo* imaging purposes, the compounds of the invention are radiolabelled. Therefore, in a further aspect of the present invention is provided a radiolabelled version of the compound of the invention. A radiolabelled compound of the invention is suitably labelled with γ -emitting radioisotopes useful for single photon emission computed tomography (SPECT) imaging. The most preferred imaging moieties are γ -emitting radiometals, in particular ^{99m}Tc. In the case of ^{99m}Tc, labelling is accomplished by the complexation of ^{99m}Tc by the chelate portion of the compound of the invention. The use of ^{99m}Tc in radiopharmaceutical applications is discussed in section B of "Technetium: chemistry and radiopharmaceutical applications", Klaus Schwochau, 2000, pp.373-423, published by Wiley-Vch Verlag GmbH.

25 A compound of formula (I) may be used for the manufacture of a medicament for the *in vivo* imaging or diagnosis of an amyloid-associated disease. Such a medicament is preferably radiolabelled and most preferably ^{99m}Tc labelled. Where the medicament is ^{99m}Tc labelled, sterile ^{99m}Tc-pertechnetate (TcO₄⁻) is obtained from a ^{99m}Tc radioisotope generator by elution with sterile saline. ^{99m}TcO₄⁻ produced from a generator is relatively unreactive, and must be reduced to a lower

30

oxidation state before use as a radiopharmaceutical. The most commonly used reducing agent is stannous chloride. Radiolabelling is carried in a mixture comprising a compound of the invention, TcO_4^- , and a suitable reducing agent by methods known to those skilled in the art (see section B of "Technetium: chemistry and radiopharmaceutical applications", Klaus Schwochau, 2000, pp.373-423, published by Wiley-Vch Verlag GmbH).

Accordingly, a further aspect of the present invention is the use of a Tc radiolabelled compound of the invention for therapeutic and diagnostic methods, for example, *in vivo* imaging and diagnosis of amyloid-associated diseases.

10 "Amyloid-associated diseases" include Alzheimer's disease, familial Alzheimer's disease, type II diabetes, Down's syndrome, homozygotes for the apolipoprotein E4 allele, rheumatoid arthritis, systemic amyloidosis (primary and secondary), and haemorrhagic stroke. Preferably, the compounds of the invention may be used to image Alzheimer's disease.

15 In the alternative, there is provided a method for *in vivo* imaging and diagnosis of amyloid-associated disease in a subject comprising administration of a radiolabelled compound of the invention. "Amyloid-associated" diseases include Alzheimer's disease, familial Alzheimer's disease, type II diabetes, Down's syndrome, homozygotes for the apolipoprotein E4 allele, rheumatoid arthritis,

20 systemic amyloidosis (primary and secondary), and haemorrhagic stroke. The method is especially preferred for *in vivo* imaging and diagnosis of Alzheimer's disease.

A radiolabelled compound according to the invention is preferably administered in a radiopharmaceutical formulation comprising the compound of the invention.

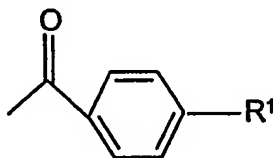
25 A "radiopharmaceutical formulation" is defined in the present invention as a formulation comprising a radiolabelled compound of the invention preferably a Tc labelled compound in a form suitable for administration to humans. Administration is preferably carried out by injection of the formulation as an aqueous solution. Such a formulation may optionally contain further ingredients such as buffers;

30 pharmaceutically acceptable solubilisers (e.g. cyclodextrins or surfactants such as Pluronic, Tween or phospholipids); pharmaceutically acceptable stabilisers or

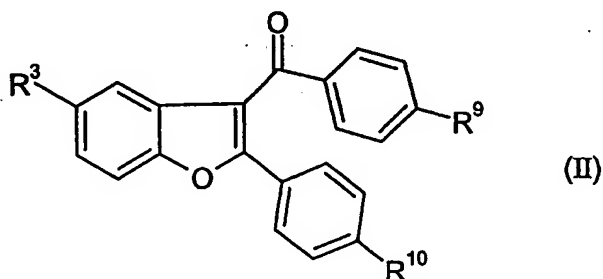
antioxidants (such as ascorbic acid, gentisic acid or para-aminobenzoic acid) or bulking agents for lyophilisation (such as sodium chloride or mannitol). The radiopharmaceutical formulation is administered in an amount which gives a reliable image, for example for a Tc-labelled compound, in an amount of 0.1 to 50mCi in about 0.5 to 5.0ml at the time of administration.

Conveniently, in a further embodiment of the invention, a kit for the preparation of ^{99m}Tc -labelled compounds for imaging and diagnosis of amyloid-associated disease is provided. A "kit" as defined by the present invention is an embodiment of the invention designed to give sterile, pyrogen-free radiopharmaceutical products suitable for human administration, e.g. via injection into the bloodstream. When the radiometal is ^{99m}Tc , the kit comprises a vial containing the compound-chelate conjugate of the invention together with a pharmaceutically acceptable reducing agent such as sodium dithionite, sodium bisulphite, ascorbic acid, formamidine sulphinic acid, stannous ion, Fe(II) or Cu(I), preferably a stannous salt such as stannous chloride or stannous tartrate. In use, a source of ^{99m}Tc is added to the vial to produce the radiolabelled compound.

Compounds of formula (I) in which R is the group



may be prepared from the corresponding compound of formula (II)



20

or a protected derivative thereof, wherein one of R^9 and R^{10} is $-\text{OCR}^4\text{R}^5\text{C}(\text{O})\text{OH}$

wherein R^4 and R^5 are as defined for the compound of formula (I);

the other is $-(O)_n-C_{1-6}$ alkyl- NR^7R^8 wherein n , R^7 , and R^8 are as defined for the compound of formula (I); and R^3 is as defined for the compound of formula (I);

by coupling with a chelate of formula



wherein [A] is as defined for the compound of formula (I).

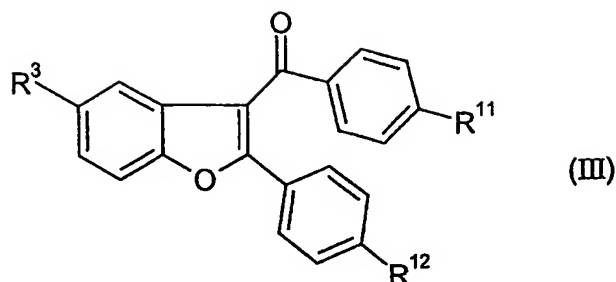
Compounds of formula (I) in which R^6 is C_{1-6} alkyl may be prepared by alkylation of the corresponding compound of formula (I) in which R^6 is hydrogen.

This coupling reaction may be performed using standard methods of amide bond
10 formation. For example, the reaction may be performed in an aprotic solvent, such as N,N-dimethyl formamide, in the presence of an organic base such as N-methylmorpholine (NMM) and 2-(1H-benzotriazole-1-yl)-1,1,3,3-tetramethyluronium tetrafluoroborate (TBTU), at a non-extreme temperature such as 10° to 40°C, preferably at ambient temperature.

15 Compounds of formula (II) are useful intermediates in the preparation of compounds of formula (I) and therefore represent a further aspect of the present invention.

Compounds of formula (II) may conveniently be prepared by hydrolysis of the corresponding ester, suitably a C_{1-6} alkyl ester, preferably the methyl ester. The
20 hydrolysis may be performed using conventional methods, suitably alkali or acid hydrolysis, for example using aqueous sodium hydroxide.

Esterified compounds of formula (II) may be prepared from the corresponding compound of formula (III)



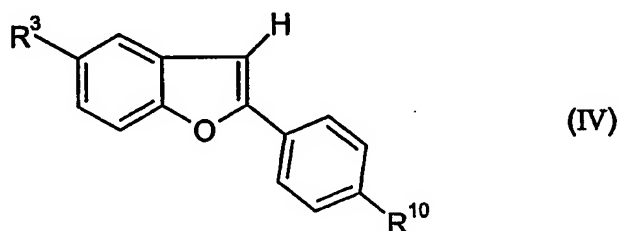
wherein one of R^{11} and R^{12} is hydroxy and the other is $-(O)_n-C_{1-6}$ alkyl- NR^7R^8 wherein n , R^7 , and R^8 are as defined for the desired compound of formula (II);

and R^3 is as defined for the compound of formula (II);

- 5 by reaction with the bromoacetate of formula $BrCR^4R^5C(O)OCH_3$ wherein R^4 and R^5 are as defined in the compound of formula (II) and are preferably both hydrogen.

Compounds of formula (III) may be prepared by methods analogous to those described in the literature (for example in Howlett et al, see above) or using the
 10 methods described in the examples below.

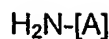
Compounds of formula (I) in which R is hydrogen, may be prepared from the corresponding compound of formula (IV):



- 15 or a protected derivative thereof, wherein R^{10} is $-OCR^4R^5C(O)OH$ as defined for the compound of formula (II); and

R^3 is as defined for the compound of formula (I);

by coupling with a chelate of formula



wherein [A] is as defined for the compound of formula (I).

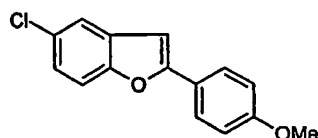
Coupling of a compound of formula (IV) with the chelate may be achieved as described above for coupling a compound of formula (II) with a chelate.

- 5 Compounds of formula (IV) may be prepared by methods analogous to those described above for preparation of compounds of formula (II) and as exemplified below.

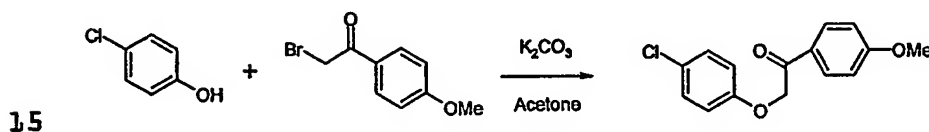
The invention may be illustrated by the following non-limiting examples.

Examples

1.0 Intermediate 1A



Step(i)



- To a solution of 4-chlorophenol (Aldrich) (2.134g, 16.6 mmol, 1 eq) and 4-methoxy bromoacetophenone (Aldrich) (4.183g, 18.26 mmol, 1.1 eq) in acetone (50 mL) was added potassium carbonate (3.516 g, 1.53eq, 25.48 mmol). The mixture was stirred and heated at reflux temperature for 5 hours. The reaction was then allowed to cool to room temperature, and filtered. The filtrate was evaporated to dryness, then dissolved in ethyl acetate (50 mL). The organic layer was then washed sequentially with 2M (NaOH)-twice, water, and saturated NaCl solution. The organic layer was then dried (Na₂SO₄), filtered and evaporated to dryness to afford a yellow solid. This was crystallised using ethyl acetate (EtOAc)/Petrol to afford the desired product as an off white crystal.
- 20
- 25

^1H NMR (CDCl_3) δH ppm 3.86 (s, 3H, OCH_3), 5.18 (s, 2H, CH_2), 6.82 (d, 2H, aromatic), 6.96 (d, 2H, aromatic), 7.21 (d, 2H, aromatic), 7.96 (d, 2H, aromatic)

5 Step(ii)

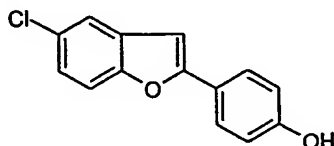


10 The compound of step (i) (3.244g, 11.72 mmol)(30) was added to a mixture of polyphosphoric acid (50g) and xylene (50 mL), and stirred using an overhead stirrer. The resultant mixture was heated at 140 C until completion (monitoring by thin-layer chromatography (TLC)). The reaction mixture was allowed to cool to room temperature (RT) and ethyl acetate (100 mL) was added. The upper organic layer was decanted off. The polyphosphoric acid was treated with water (100 mL) and eliminated from the flask. A further 200 mL water was added, and the aqueous phase was extracted with 3x100 mL EtOAc. The combined organic layer were then washed with brine (100 mL), dried (Na_2SO_4), filtered and evaporated to afford a crude solid. This was purified by recrystallisation using ethyl acetate.

20

^1H NMR (CDCl_3) δH ppm 3.84 (s, 3H, OCH_3), 6.80 (s, 1H, aromatic), 6.95 (d, 2H, aromatic), 7.18 (d, 1H, aromatic), 7.39 (d, 2H, aromatic), 7.49 (d, 1H, aromatic), 7.77 (d, 2H, aromatic).

25 Intermediate 1



A solution of Intermediate 1A (100 mg, 0.38 mmol)(21) under nitrogen, in dichloromethane (DCM)(7 mL) was cooled in an ice bath. Boron tribromide solution (1M in DCM, 3.8 mL, 3.8 mmol, 10 eq) was added slowly. The dark brown reaction mixture was stirred under N₂, and allowed to warm slowly to room
5 temperature. After 4 hours, TLC indicated that reaction was complete (eluant : 5:1 petrol ether : ethyl acetate). The reaction was quenched by addition of 20 mL 10% HCl solution, and was diluted with DCM.

The organic layer was separated off, and the aqueous phase was extracted with ethyl acetate. The organic layers were dried (Na₂SO₄), filtered and evaporated to
10 dryness to afford a dirty brown/black solid. This was purified by flash column chromatography (eluent 5:1 petrol : ethyl acetate) to afford a light brown solid.

¹H NMR (CDCl₃) δH ppm 6.79 (s, 1H, aromatic), 6.87 (d, 2H, aromatic), 7.18 (d, 1H, aromatic), 7.37 (d, 1H, aromatic), 7.49 (d, 1H, aromatic), 7.73 (d, 2H,
15 aromatic).

Intermediate 2 diethyl(3-chloropropyl)amine



20

A solution of 3-diethylamino-1-propanol (Aldrich) (10g, 76mmol) in DCM was stirred at 0°C. A solution of thionyl chloride (DCM 30mL) was then added drop-wise such that the reaction temperature remained at 0°C. Once addition is complete, the reaction was refluxed for one hour. The reaction was allowed to
25 cool and the excess DCM and thionyl chloride were removed under vacuum. The residue was basified (pH8-9) using 2N NaOH and the aqueous mixture extracted with DCM (3 x 100mL). The organic extracts were combined, dried (MgSO₄) and evaporated to give a brown oil. The oil was distilled (160°-165°C) to give the product as a clear oil (4.2g 36.9%).

30

^1H NMR (CDCl_3) δH ppm 1.02 (6H, $2\times\text{CH}_3$), 1.89 (2H, CH_2), 2.53 (6H, $3\times\text{CH}_2$), 3.59 (2H, CH_2)

m/z 150/151 $\text{M}+\text{H}$

5 Intermediate 3 diethyl-chloroethylamine



10 A solution of 2-diethylaminoethanol (Lancaster) (10g, 85mmol) in DCM was stirred at 0°C. A solution of thionyl chloride (30mL DCM) was added such that the temperature of the reaction remained at 0°C. Once addition is complete, the reaction is refluxed for one hour. The reaction is allowed to cool and the excess DCM and thionyl chloride were removed under vacuum. The residue was basified (pH8-9) using 2N NaOH and the aqueous mixture extracted with DCM (3 x 15 100mL). The organic extracts were combined, dried (MgSO_4) and evaporated to give a brown oil. The oil was distilled (165°-169°C) to give the product as a clear oil (6.7g 58%).

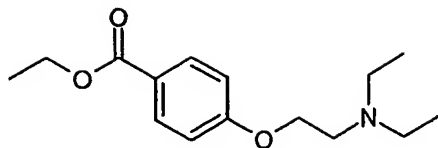
20 ^1H NMR (CDCl_3) δH ppm 1.02 (6H, $2\times\text{CH}_3$), 2.58 (4H, $2\times\text{CH}_2$), 2.78 (2H, CH_2) 3.52 (2H, CH_2)

m/z 136/137 $\text{M}+\text{H}$

Example 1

25

Example 1(i) ethyl-4-(diethylaminoethoxy) benzoate

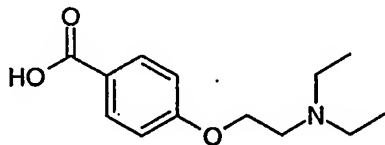


The title compound was prepared by alkylation of ethyl-4-hydroxybenzoate with diethylchloroethylamine using methods analogous to those described in Example

5 2(i).

^1H NMR (CDCl_3) δH ppm 1.07 (6H, $2\times\text{CH}_3$), 1.38 (3H, CH_3), 2.63 (4H), 2.89 (2H, CH_2), 4.1 (2H, OCH_2) 4.35 (2H, OCH_2), 6.91 (2H), 7.97 (2H).

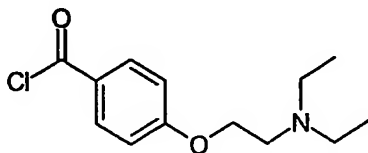
10 Example 1(ii) 4-(3-diethylaminoethoxy) benzoic acid



The title compound was prepared from the compound of Example 1(i) using methods analogous to those described in Example 2(ii) to give the expected
15 product 5.2g, 91% yield.

^1H NMR ($d_6\text{DMSO}$) δH ppm 1.22 (6H, $2\times\text{CH}_3$), 3.18 (4H), 3.49 (2H, CH_2), 4.46 (2H, OCH_2), 7.05 (2H), 7.89 (2H).

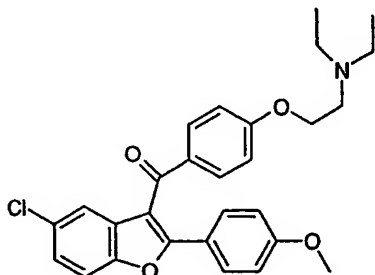
20 Example 1(iii) 4-(2-diethylaminoethoxy) benzoyl chloride



The title compound was prepared from the compound of Example 1(ii) using
25 methods analogous to those in Example 2(iii), to give 5.4g, 100% yield.

^1H NMR (d^6DMSO) δH ppm 1.27 (6H, $2\times\text{CH}_3$), 3.17 (4H), 3.49 (2H) 4.47 (2H, OCH_2), 7.07 (2H), 7.9 (2H).

- 5 Example 1(iv) 5-chloro-2-(4-methoxyphenyl)-3-[4-(2-diethylaminoethoxy)benzoyl]benzofuran.

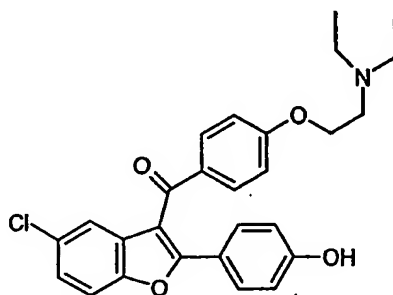


- 10 The methods of Example 2(iv) were used to couple Intermediate 1A to the compound of Example 1(iii) to give the expected product (1.21g, 68%)

^1H NMR (CDCl_3) δH ppm 1.28 (6H, $2\times\text{CH}_3$), 2.99 (4H), 3.22 (2H), 3.81 (3H, CH_3), 4.38 (2H, OCH_2), 6.84, (4H), 7.27 (1H), 7.46 (2H), 7.63 (2H), 7.83 (2H).

15

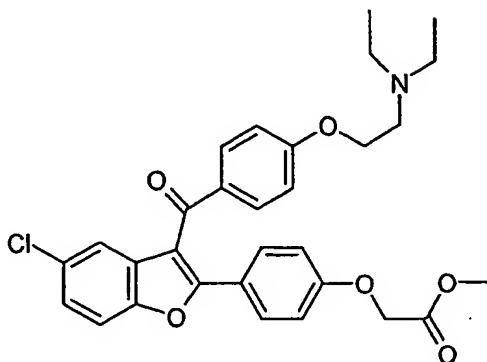
- Example 1(v) 5-chloro-2-(4-hydroxyphenyl)-3-[4-(3-diethylaminoethoxy)benzoyl]benzofuran.



- 20 The title compound was prepared from the compound of Example 1(iv) using methods analogous to those described in Example 2(v).

^1H NMR (DMSO) δ H ppm 0.91 (6H, 2xCH₃), 2.5 (4H), 2.27 (2H), 4.03 (2H, OCH₂), 6.96 (2H), 6.87 (2H), 7.35 (4H), 7.68 (3H).

Example 1(vi)

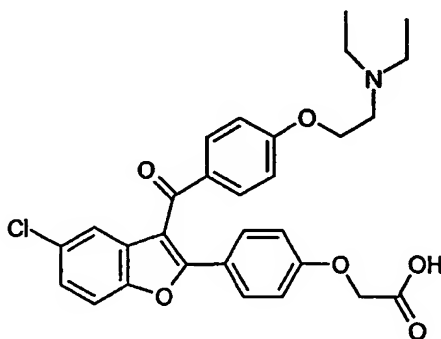


5

The title compound was prepared from the compound of Example 1(v) using methods analogous to those described in Example 4(iv).

10 ^1H NMR (DMSO) δ H ppm 0.91 (6H, 2xCH₃), 2.6 (4H), 2.88 (2H), 3.8 (3H), 4.03 (2H, OCH₂), 4.67 (2H), 6.84 (4H), 7.4 (1H), 7.49 (2H), 7.61 (2H), 7.84 (2H).

Example 1(vii)



15

The title product was prepared from the corresponding methyl ester using the following route:

A solution of the compound of Example 1(vi) in methanol was stirred at RT. 2N

NaOH (equivalent volume) was added and the reactions were stirred at RT for 2 hours. The reaction was diluted with H₂O and acidified to pH4. The aqueous was extracted with DCM (3x) and the organic extracts were combined, dried and evaporated to give the title compound.

5

¹H NMR (CDCl₃) δH ppm 1.36 (6H, 2xCH₃), 3.25 (6H), 4.4 (2H, OCH₂), 4.5 (2H), 6.62 (2H), 6.68 (2H), 7.14 (2H), 7.34 (1H), 7.46 (1H), 7.53 (2H), 8.1 (1H).

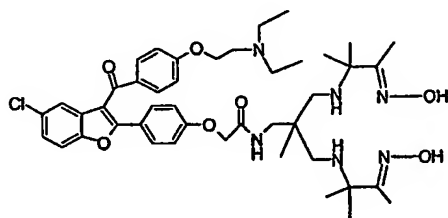
m/z 522.26/524.49

10

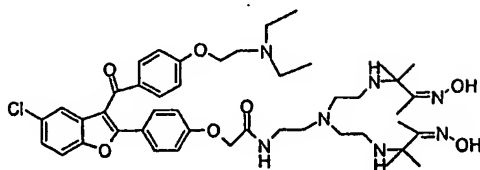
Example 1(viii) Coupling to chelate

The benzofuran of Example 1(vii) (1 eq) and chelate (as a free primary amine)(1 eq) are dissolved in dimethylformamide. TBTU (2-(1H-benzotriazole-1-yl)-1,1,3,3,-
15 tetramethyluroniumtetrafluoroborate) (1.2 eq) and N-methylmorpholine (1.2 eq) are then added. The reaction is stirred at room temperature for 12 hours. The final product is purified by preparative HPLC (PRP column) without any further treatments.

20 Pn44 conjugated (from Example 1(vii))

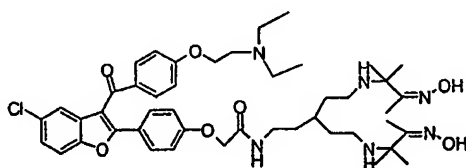


M/S : ES⁺ : m/z 819.3, 410.6 ; M+H, (M+2)/2

Pn216 conjugated (from Example 1(vii))

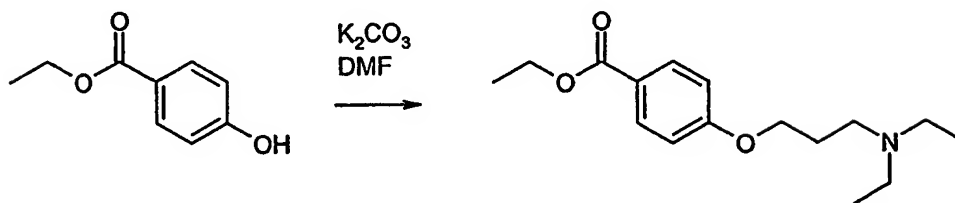
M/S : ES+ : m/z 848.2, 425.1 ; M+H, (M+2)/2

5

cPn216 conjugated (from Example 1(vii))

M/S : ES+ : m/z 847.3, 424.6 ; M+H, (M+2)/2

10

Example 2Example 2(i) ethyl-4-(diethylaminopropoxy) benzoate

15

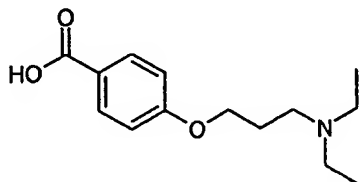
A solution of ethyl-4-hydroxybenzoate (10g, 6mmol), Intermediate 2 (10g, 6.7mmol) and potassium carbonate (16.6g, 12mmol) in DMF was heated at 100°C for 48 hours. On completion, the reaction was poured onto 200mL ice-water and the mixture extracted using ethyl acetate (3 x 100mL). The organic extracts were combined, dried (Na₂SO₄) and evaporated. The residue was purified by flash chromatography to give the title compound 7.6g, 61% as a pale orange oil.

20

¹H NMR (CDCl₃) δH ppm 1.07 (6H, 2xCH₃), 1.35 (3H, CH₃), 1.92 (2H, CH₂), 2.56 (6H), 4.06 (2H, OCH₂) 4.35 (2H, OCH₂), 6.91 (2H), 7.97 (2H).

Example 2(ii) 4-(3-diethylaminopropyloxy) benzoic acid

5



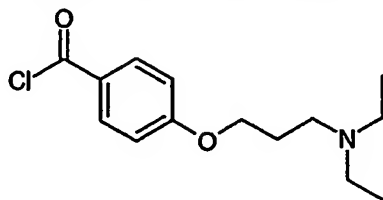
A solution of the compound from Example 2(i) (8g, 29mmol) in H₂O (50mL) was stirred at RT. Conc. HCl was added (50mL) and the reaction was heated at reflux for 18 hours. The reaction was allowed to cool and evaporated to dryness. The solid was dissolved in hot ethanol and petroleum ether was added to afford the expected product as white platelets, 72g, 88%.

10

¹H NMR (d⁶DMSO) δH ppm 1.19 (6H, 2xCH₃), 2.17 (2H), 3.12 (6H, CH₂), 4.14 (2H, OCH₂), 7.02 (2H), 7.88 (2H).

15

Example 2(iii) 4-(diethylaminopropyloxy) benzoyl chloride.



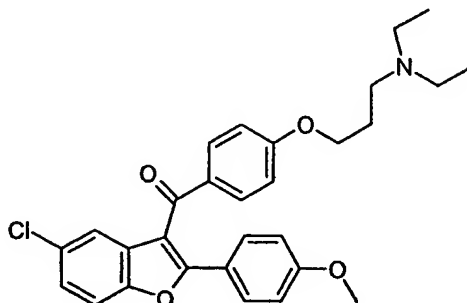
A solution of the compound from Example 2(ii)(6g, 24mmol) in DCM was stirred at 0°C. Thionyl chloride was added drop-wise and the reaction was heated at reflux for 24hrs. The reaction was allowed to cool and the solvents removed under vacuum to afford the expected compound as a pungent crude solid. 5.9g, 100%. The solid was used without further purification.

20

¹H NMR (d⁶DMSO) δH ppm 1.19 (6H, 2xCH₃), 2.16 (2H), 3.12 (6H) 4.12 (2H, OCH₂), 7.03 (2H), 7.89 (2H).

25

Example 2(iv) 5-chloro-2-(4-methoxyphenyl)-3-[4-(3-diethylaminopropoxy)benzoyl]benzofuran.



5

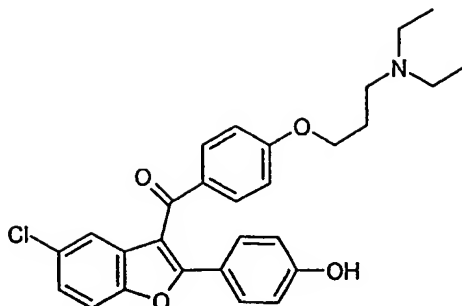
A solution of 5-chloro-2-(4-methoxyphenyl)benzofuran (1g, 3.9mmol) in DCM (100mL) was stirred at 0°C under nitrogen. The compound of Example 2(iii) (1.68g, 6.9mmol) was added followed by the drop-wise addition of tinIV chloride solution (1M DCM, 11.6mmol). The colour became a deep red and the reaction was stirred at RT. for 3hrs. The resulting suspension was poured onto ice-water (250mL) and the organic portion separated. The aqueous layer was extracted with DCM and the organic portions were combined, dried (Na₂SO₄) and evaporated. The residue was purified by flash chromatography to give the title compound as a yellow tar (1.34g, 72%).

15

¹H NMR (CDCl₃) δH ppm 1.38 (6H, 2xCH₃), 2.36 (2H), 3.12 (4H), 3.81 (3H, CH₃), 4.12 (2H, OCH₂), 6.83, (4H), 7.29 (1H), 7.46 (2H), 7.65 (2H), 7.82 (2H).

Example 2(v) 5-chloro-3-[4(3-diethylaminopropoxy)benzoyl]-2-(4-hydroxyphenyl)benzofuran

20



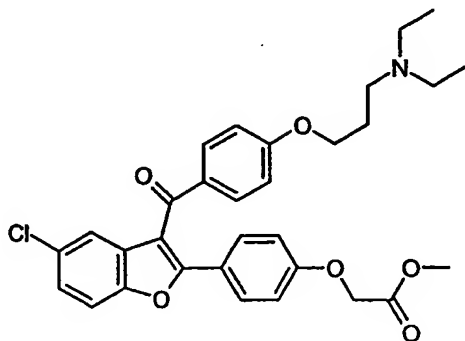
A solution of the compound of Example 2(iv) (4.3g, 8.7mmol) in dichloroethane (100mL) was cooled to 10°C. Aluminium chloride (50mmol) was added followed by 2-methyl-5-tertbutyl thiophenol (17.4mmol) and the reaction was heated at 40°C for 4 hours. Methanol (20mL) was added followed by H₂O (150mL). The organic phase was separated, concentrated under vacuum and purified by flash chromatography to give the expected compound as a bright yellow oil. 1.9g, 46% yield.

10

¹H NMR (CDCl₃) δH ppm 1.34 (6H, 2xCH₃), 2.18 (2H), 3.08 (6H), 4.05 (2H, OCH₂), 6.66 (2H), 6.79 (2H), 7.25 (3H), 7.43 (1H), 7.60 (2H), 7.75 (1H).

Example 2(vi)

15



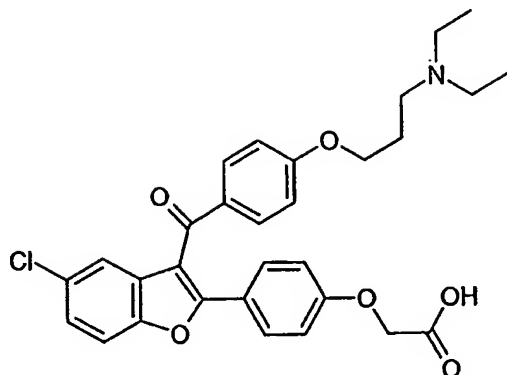
The title compound was prepared from the compound of Example 2(v) using methods analogous to those described in Example 4(iv).

20

¹H NMR (CDCl₃) δH ppm 0.18 (6H, 2xCH₃), 2.1 (2H), 2.8 (6H), 3.8 (3H) 4.07

(2H), 4.63 (2H, OCH₂), 6.84 (4H), 7.26 (2H), 7.45 (1H), 7.64 (2H), 7.83 (2H).

Example 2(vii)



- 5 The title compound was prepared using methods analogous to those described in Example 1(vii).

δ H ppm 1.12 (6H, 2xCH₃), 2.07 (2H), 6.76 (6H), 4.22 (2H, OCH₂), 4.5 (2H), 6.85 (2H), 6.94 (2H), 7.5 (3H), 7.65 (1H), 7.76 (2H), 7.82 (1H).

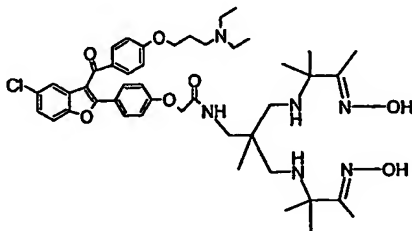
10

m/z 536.34/538.37

Example 2(viii)

- 15 The benzofuran of Example 2(vii) was coupled to various chelates using methods analogous to those described in Example 1(viii).

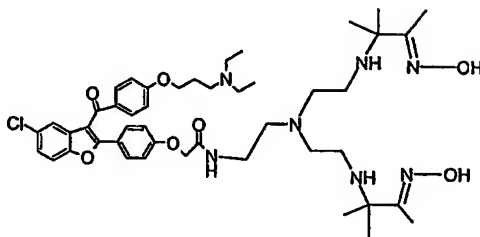
Pn44 conjugated (from Example 2(vii))



20

M/S : ES+ : m/z 833.3, 417.5, M+H, (M+2)/2

Pn216 conjugated (from Example 2(vii))

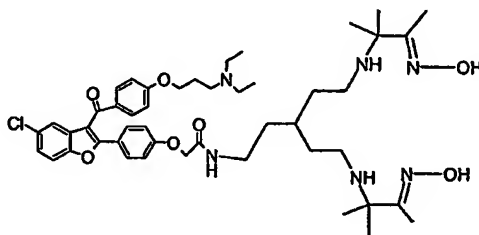


5

M/S : ES+ : m/z 862.3, 432.1, M+H, (M+2)/2

cPn216 conjugated (from Example 2(vii))

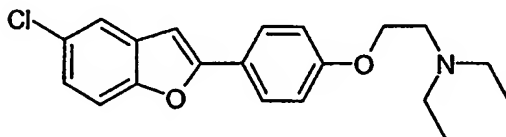
10



M/S : ES+ : m/z 861.3, 431.7, M+H, (M+2)/2

Example 3

15 Example 3(i) 5-chloro-2-[4(2-diethylaminoethoxy)phenyl]benzofuran



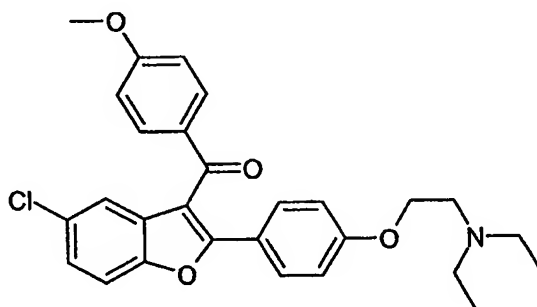
A solution of Intermediate 1 (5g, 20mmol) in DMF was stirred at RT. Intermediate
20 3 (22mmol) was added followed by potassium carbonate. The reaction was
heated to 90°C for 2 hours. The reaction was allowed to cool and was then

poured onto ice-water. The resulting solid was collected by filtration and dried to give the title compound as a white solid 4.5g, 65.5%).

¹H NMR (CDCl₃) δH ppm 1.08 (6H, 2xCH₃), 2.65 (2H, 2xCH₂), 2.91 (4H, 2xCH₂),
5 4.09 (2H, OCH₂) 6.8 (1H), 6.97 (2H), 7.18 (1H), 7.38 (1H), 7.49 (1H), 7.76 (2H)

m/z 342/344 M+H

Example 3(ii)



10

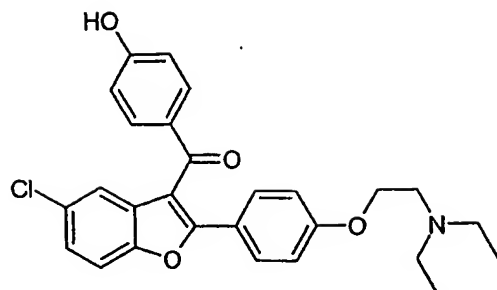
A solution of the compound of Example 3(i) (3.5g, 10mmol) and p-anisoylchloride (18mmol) in dry DCM was stirred at 0°C under nitrogen. A solution of SnCl₄ (1M DCM, 30mmol) was added drop-wise and stirring was continued for 2 hours at RT.
15 under nitrogen. The reaction mixture was added carefully to 100mL of ice-water and the aqueous was extracted with ethylacetate (3x100mL). The organic extracts were combined, dried (Na₂SO₄) and evaporated to give a bright yellow oil. This was purified by flash chromatography (silica flash) to give the title compound as a white solid. (2.1g 45%).

20

¹H NMR (CDCl₃) δH ppm 1.45 (6H, 2xCH₃), 3.24 (4H, 2xCH₂), 3.43 (2H, CH₂),
3.87 (3H, CH₃) 4.55 (2H, OCH₂) 6.85 (4H), 7.3 (1H), 7.44 (2H), 7.67 (2H), 7.86 (2H).

25 m/z 477/479 M+H

Example 3(iii)

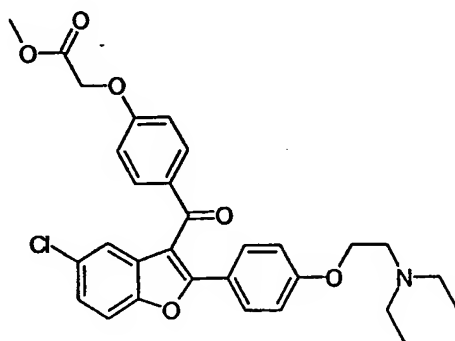


Pyridine hydrochloride (10g) was heated to 170°C. Solid product from Example 3(ii) was added and the reaction was heated at 170°C for a further 2 hours. The reaction was allowed to cool and ice-water added (100mL) and the aqueous mixture was extracted using DCM (3x100mL). The organic extracts were combined, dried (Na₂SO₄) and evaporated giving an oily residue. This was purified by flash chromatography (SiO₂) to give the title compound as a pale yellow foam (0.8g, 63%)

¹H NMR (CDCl₃) δH ppm 0.91 (6H, 2xCH₃), 2.5 (4H, 2xCH₂), 2.75 (2H, CH₂), 4.03 (2H, OCH₂), 6.71 (2H), 6.88 (2H), 7.37 (4H), 7.69 (2H).

m/z 451/453 M+H

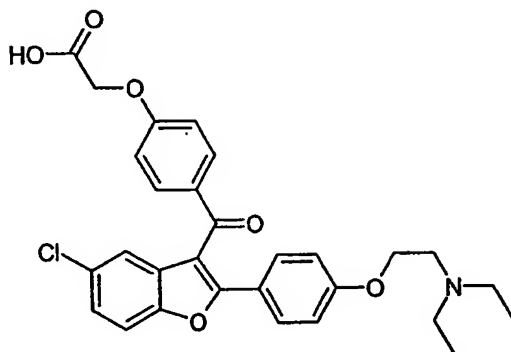
Example 3(iv)



The title compound was prepared from the compound of Example 3(iii) using methods analogous to those described in Example 4(iv). 0.21g 37% of product were obtained.

¹H NMR (CDCl₃) δH ppm 1.06 (6H, 2xCH₃), 2.62 (4H, 2xCH₂), 2.84 (2H, CH₂), 3.8 (3H, CH₃), 4.04 (2H, OCH₂) 4.6 (2H, OCH₂), 6.83 (4H), 7.29 (1H), 7.47 (2H), 7.61 (2H), 7.83 (2H).

5 Example 3(v)



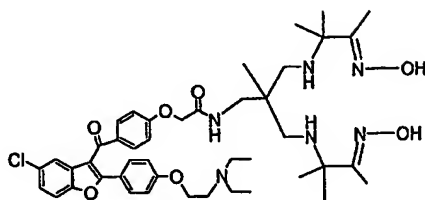
The title compound was prepared using methods analogous to those described
10 in Example 1(vii).

¹H NMR (DMSO) δH ppm 1.14 (6H, 2xCH₃), 3.04 (4H), 3.3 (2H), 4.3 (2H, OCH₂),
4.6 (2H), 6.76 (2H), 6.90 (2H), 7.4 (3H), 7.6 (3H), 7.7 (1H).

15 m/z 522.28/524.47

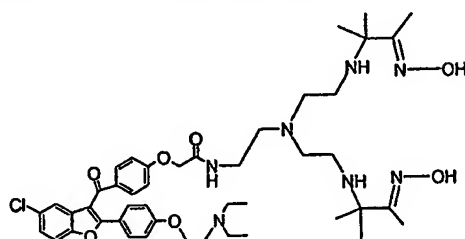
Example 3(vi)

The compound of Example 3(v) was conjugated to a variety of chelates using
20 methods analogous to those described in Example (viii)

Pn44 conjugated (from Example 3(v))

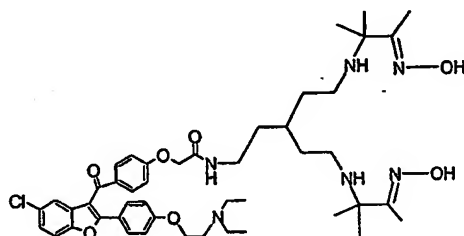
M/S : ES+ : m/z 819.3, 410.2 ; M+H, (M+2)/2

5

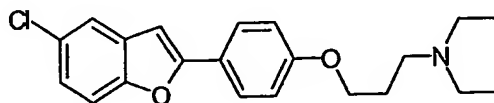
Pn216 conjugated (from Example 3(v))

M/S : ES+ : m/z 848.3, 424.7 ; M+H, (M+2)/2

10

cPn216 conjugated (from Example 3(v))

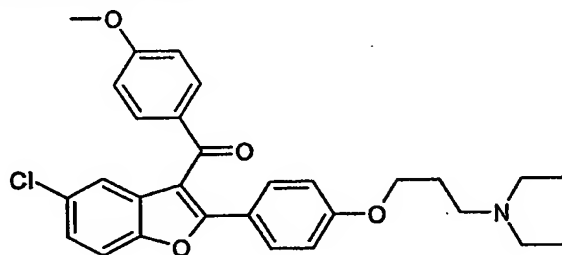
M/S : ES+ : m/z 847.3, 424.5 ; M+H, (M+2)/2

15 Example 4Example 4(i) 5-chloro-2-[4-(3-diethylaminopropoxy)phenyl]benzofuran

A solution of the Intermediate 1 (4g, 16mmol) in DMF was stirred at RT. Intermediate 2 (18mmol) was added followed by potassium carbonate. The reaction was heated to 90°C for 2 hours. The reaction was allowed to cool and
 5 was then poured onto ice-water. The resulting solid was collected by filtration and dried to give the title compound as a white solid 4.2g, 76.4%).

¹H NMR (CDCl₃) δH ppm 1.04 (6H, 2xCH₃), 1.94 (2H, CH₂), 2.58 (6H, 3xCH₂), 4.07 (2H, OCH₂) 6.79 (1H), 6.95 (2H), 7.17 (1H), 7.38 (1H), 7.49 (1H), 7.74 (2H)
 10 m/z 357/359 M+H

Example 4(ii) 5-chloro-2-[4-(3-diethylaminopropoxy)phenyl]-3-(4-methoxybenzoyl)benzofuran



15

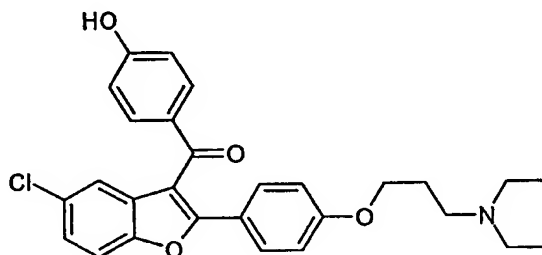
A solution of the product from Example 4 (i) (1g, 2.8mmol) in dry DCM was stirred at 0°C under nitrogen. Anisoyl chloride (4.9mmol) was added followed by a solution of SnCl₄ (1M DCM, 4.9mmol). An immediate red colour occurred and stirring was continued for 2 hours. The reaction mixture was added carefully to
 20 100ml of ice-water and the aqueous was extracted with ethylacetate (3x100ml). The organic extracts were combined, dried (Na₂SO₄) and evaporated to give a bright yellow oil. This was purified by flash chromatography (silica flash) to give the title compound as a pale yellow solid. (0.8g 58%).

25 ¹H NMR (CDCl₃) δH ppm 1.42 (6H, 2XCH₃), 2.37 (2H, CH₂), 3.17 (6H, 3XCH₂), 3.89 (3H, CH₃) 4.08 (2H, OCH₂) 6.82 (4H), 7.26 (1H), 7.45 (2H), 7.63 (2H), 7.84 (2H).

m/z 491/493 M+H

Example 4(iii) 5-chloro-2-[4-(3-diethylaminopropoxy)phenyl]-3-(4-hydroxybenzoyl)benzofuran

5

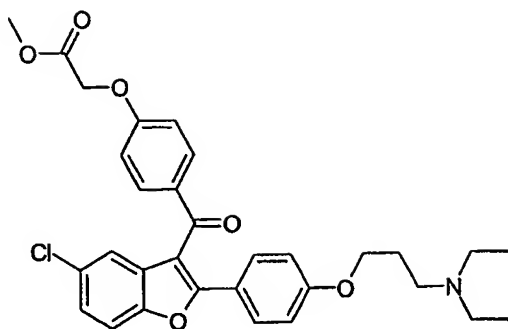


Pyridine hydrochloride (10g) was heated to 170°C. Solid product from Example 4(ii) was added and the reaction was heated at 170°C for a further 2 hours. The reaction was allowed to cool and ice-water added (100mL) and the aqueous mixture was extracted using DCM (3x100mL). The organic extracts were combined, dried (Na₂SO₄) and evaporated giving an oily residue. This was purified by flash chromatography (SiO₂) to give the title compound as a pale yellow solid (1.2g, 61.4%)

15

¹H NMR (CDCl₃) δH ppm 1.11 (6H, 2xCH₃), 2.01 (2H, CH₂), 2.94 (4H, 2xCH₂), 3.31 (2H, CH₂) 4.08 (2H, OCH₂) 6.76 (2H), 6.96 (2H), 7.4 (2H), 7.56 (2H), 7.68 (2H), 7.75 (1H).

20 m/z 477/479 M+H

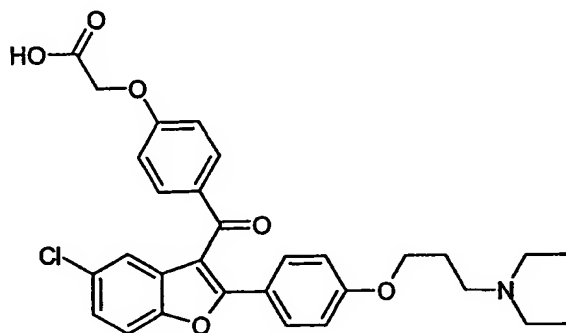
Example 4(iv)

- 5 A solution of the product from Example 4(iii) (0.8g, 1.8mmol) in dry DMF was stirred at RT. Potassium carbonate was added and the reaction was heated to 90°C for 2 hours. The reaction mixture was poured onto ice-water and the aqueous extracted using DCM (3x100mL). The organic extracts were combined, dried (Na₂SO₄) and evaporated to give a yellow residue. This was purified using
- 10 flash chromatography to give the title compound as an orange tar (0.13g, 21%).

¹H NMR (CDCl₃) δH ppm 1.18 (6H, 2xCH₃), 2.22 (2H, CH₂), 2.79 (6H, 3xCH₂), 3.8 (3H, CH₃) 4.08 (2H, OCH₂) 4.67 (2H, CH₂) 6.83 (4H), 7.27 (1H), 7.45 (2H), 7.65 (2H), 7.83 (2H).

15

m/z 549/551 M+H

Example 4 (v)

20

The title compound was prepared using methods analogous to those described in Example 1(vii).

¹H NMR (DMSO) δH ppm 1.26 (6H, 2xCH₃), 2.15 (2H), 3.22 (6H), 3.38 (2H), 4.16
5 (2H, OCH₂), 4.79 (2H), 6.9 (2H), 7.0 (2H), 7.5 (2H), 7.66 (2H), 7.84 (3H).

m/z 536.32/538.43

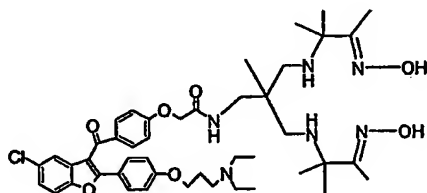
Example 4(vi)

10

The benzofuran of Example 4(v) was conjugated to a variety of chelates using methods analogous to those described in Example 1(viii)

Pn44 conjugated (from Example 4(v))

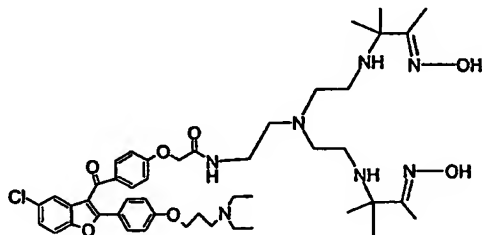
15



M/S : ES⁺ : m/z 833.3, 417.6, M+H, (M+2)/2

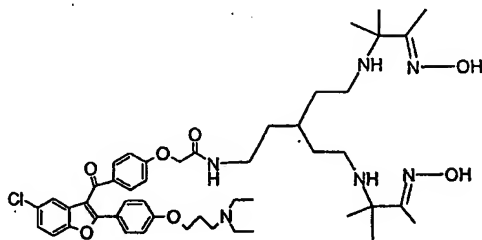
Pn216 benzofuran3 conjugated (from Example 4(v))

20



M/S : ES⁺ : m/z 862.3, 432.1, M+H, (M+2)/2

cPn216 benzofuran3 conjugated (from Example 4(v))



5 M/S : ES+ : m/z 861.3, 431.4, M+H, (M+2)/2

Example 5 ^{99m}Tc labelling of compounds of the invention (I)

A 0.1ml aliquot of a 1mg/ml solution of a compound of the invention dissolved in methanol or H₂O is transferred to a nitrogen-filled 10ml glass vial together with deoxygenated saline (0.9% w/v, 1ml) and 0.035ml aqueous NaOH (0.1M). To this solution is added technetium generator eluate (1ml, approx. 0.4GBq) and then aqueous stannous chloride solution (0.1ml, ca.10µg). The labelling pH is 9.0-10.0. Vials are incubated at ambient laboratory temperature (15-25°C) for 30 minutes to effect labelling. HPLC purification is performed to remove unlabelled starting material and radioactive impurities prior to testing. After purification the organic solvent is removed under vacuum and the sample is redissolved in about 5ml 0.1M phosphate buffer pH7.4 to give a working concentration of 6-9MBq/ml. Radiochemical purity is assessed before use by the thin layer chromatography (TLC) system described below:

- 20 i) ITLC SG 2cm x 20cm eluted with 0.9% w/v saline
- ii) Whatman No.1 2cm x 20cm eluted with 50 : 50 v/v acetonitrile : H₂O

The labelled substrates remain at, or close to, the origin in TLC system (i) and move close to the solvent front in system (ii). When analysed by appropriate detection equipment the radiochemical purity is typically in excess of 85% labelled compound.

Example 6 ^{99m}Tc labelling of compounds of the invention (II)

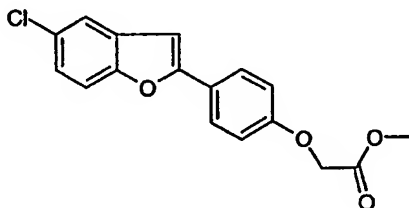
A gluconate kit, comprising of sodium gluconate (1mg) sodium bicarbonate (2 mg), stannous chloride (15 μg) is reconstituted with technetium generator eluate (5ml, 2GBq) and allowed to incubate at room temperature for 15 minutes to effect labelling. An aliquot (0.1ml) of a compound of the invention freshly dissolved in methanol (5mg/ml) is transferred to a nitrogen-filled 10ml glass vial together with 0.025 ml aqueous NaOH (0.1M) and 2ml of the ^{99m}Tc -gluconate solution. The labelling pH is 9.0. Vials are incubated at room temperature for 30 minutes to effect labelling. Purification and assessment of radiochemical purity are carried out as for Example 5.

Example 7 ^{99m}Tc labelling of compounds of the invention (III)

A 0.1ml aliquot of a compound of the invention dissolved in methanol (1mg/ml) is transferred to nitrogen-filled 10ml glass vial together with tricine dissolved in water (0.5ml, 37.5mg) and phosphinedynetrise(benzene sulphonic acid)tris sodium salt dissolved in water (0.1ml, 1mg). To this solution is added technetium generator eluate (1ml, approx 0.4GBq) and then a solution of stannous chloride in 0.1M HCl (0.02ml, ca 2 μg). The labelling pH is 4.5-5.5. Vials were incubated at 60°C for 30 minutes to effect labelling. Purification and assessment of radiochemical purity is carried out as in Example 5.

Example 8 Preparation of 3-H compounds

Example 8(i)



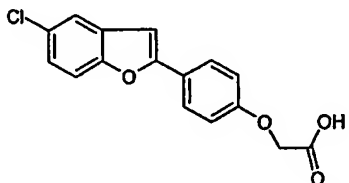
25

Intermediate 1 (200 mg, 8.17 10^{-4} mol), methylbromoacetate (0.38 mL, 4.09 $\times 10^{-3}$

mol, 5 eq) and potassium carbonate (1.129 g, 8.17x10⁻³ mol, 10 eq) were stirred in acetone (30 mL) and heated to reflux temperature. The mixture was refluxed for 5 hours. The reaction mixture was evaporated to dryness and the residue was partitioned between water and ethyl acetate. The layers were separated and the organic phase was washed with a further 2 portions of water. The organic phase was dried (Na₂SO₄) filtered and evaporated. The residue was purified by recrystallisation from ethyl acetate to afford the product as a white crystals (36)(125 mg, 48%)

¹H NMR (CDCl₃) δH ppm 3.83 (3H, OCH₃), 4.7 (2H, OCH₂) 6.84 (1H), 7.0 (d, 2H), 7.2 (d, 1H), 7.42 (d, 1H), 7.51 (d, 1H), 7.80 (d, 2H).

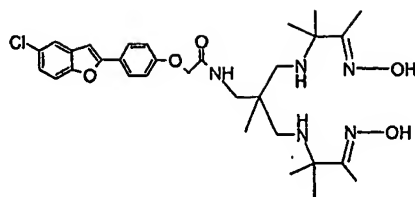
Example 8(ii)



The title compound was prepared using methods analogous to those described in Example 1(vii).

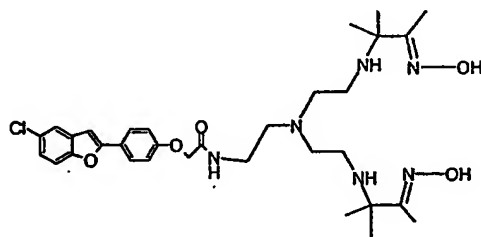
Example 8(iii)

The following chelated compounds were prepared from the compound of Example 8(ii) using methods analogous to those described in Example 1(viii). When labelled with ^{99m}Tc using the method described in Examples 5 to 7, these compounds may also be useful for imaging Alzheimer's disease and therefore represent a further aspect of the present invention.

Pn44 conjugated

M/S : ES+ : m/z: 600.2

- 5 ¹H NMR (DMSO) : d 0.718 (s, 3H); 1.09 (s, 12H), 1.67 (s, 6H), 2.0 (m, 4H); 3.09 (m, 2H); 4.59 (s, 2H); 7.08 (d, 2H); 7.27 (s, 1H); 7.31 (m, 1H); 7.63 (d, 1H); 7.69 (m, 1H); 7.87 (d, 2H); 8.21 (bt, 1H); 10.36 (bs, 1H)

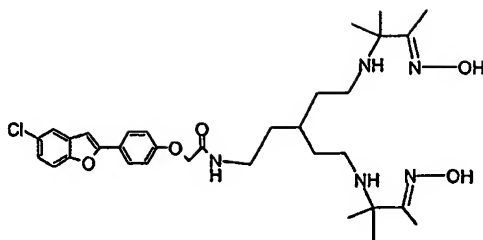
Pn216 conjugated

10

M/S : ES+ : m/z: 629.2

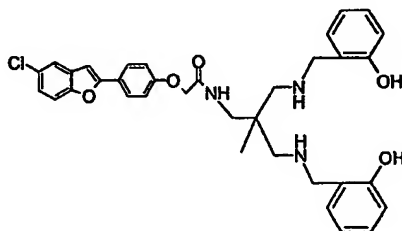
- ¹H NMR (DMSO) : 1.010 (s, 12H), 1.69 (s, 6H), 2.26 (m, 4H); 2.42 (m, 2H); 3.18 (m, 2H); 4.85 (s, 2H); 7.08 (d, 2H); 7.26 (s, 1H); 7.30 (m, 1H); 7.63 (d, 1H); 7.67 (m, 1H); 7.87 (d, 2H); 8.01 (bt, 1H); 10.38 (bs, 1H)

15

cPn216 conjugated

M/S : ES⁺ : m/z 628.2

1H NMR (DMSO) : 1.07 (s, 12H), 1.315 (m, 7H) ; 1.90 (s, 6H), 2.14 (m, 4H); 3.09
 5 (m, 2H) ; 4.48 (s, 2H); 7.03 (d, 2H); 7.20 (s, 1H); 7.26 (dd, 1H); 7.54 (d, 1H); 7.61
 (d, 1H); 7.80 (d, 2H); 8.12 (bt, 1H); 10.38 (bs, 1H)

DADP conjugated

10

M/S : ES⁺ : m/z 614.1

1H NMR : (DMSO) : 0.79 (s, 3H), 2.33 (m, 4H); 3.11 (m, 2H) ; 3.70 (s, 4H); 4.61
 (s, 2H); 4.86 (s, 1H); 6.70 (m, 4H); 7.05 (m, 6H); 7.27 (m, 2H); 7.62 (d, 1H); 7.68
 (d, 1H); 7.90 (d, 2H); 8.20 (bt, 1H), 7.94 (s, 1H)

15

Biological DataDetermination of amyloid binding

The binding of Compounds was determined compared to the ability of ¹²⁵I-beta
 amyloid protein 1-40 (¹²⁵I-BAP 1-40, Amersham Biosciences IM294)) to bind to
 20 amyloid 1-40 fibrils. Amyloid binding was essentially performed as follows.

Three fresh buffer stocks were prepared for experiments: Buffer 1, 50 mM

HEPES/0.1% Bovine Serum Albumin (BSA) pH 7.5; Buffer 2, 50 mM HEPES/0.1% BSA/400 μ M ZnCl₂ pH 7.5; Buffer 3, 50 mM HEPES/0.1% BSA/100 μ M ZnCl₂ pH 7.5.

Streptavidin coated scintillation proximity assay beads (SA-SPA beads, Amersham Biosciences) were used to immobilise fibrillar Beta-Amyloid Protein (BAP 1-40). Amyloid coated beads (SPA-BAP) were prepared by incubating 250 μ l SA-SPA beads (100 mg/ml) with 250 μ l Buffer 2, 425 μ l Buffer 1, 50 μ l biotinylated BAP 1-40 (0.5 mg/ml, Biosource 03-243), 25 μ l BAP 1-40 (10 mg/ml, Biosource 03-138). Non-specific binding SPA beads (SPA-NSB) were prepared to assess the binding of compounds to SPA beads with no associated BAP 1-40 fibrils in the following incubation: 250 μ l SA-SPA beads (100 mg/ml) with 250 μ l Buffer 2, 500 μ l Buffer 1.

SPA-BAP and SPA-NSB incubations were left for 24 hours at room temperature and then spun 1.5 ml tubes (Eppendorf, Merk, 306/0421/12) for 2 minutes at 1000 x g. The supernatants were removed and the beads were washed twice by resuspending them in 1 ml Buffer 3 followed by centrifugation for 2 minutes at 1000 x g. Finally, washed SPA-BAP and SPA-NSB beads were resuspended in 1 ml Buffer 3.

Amyloid binding of ¹²⁵I-BAP 1-40 and test compounds was performed in triplicate in 0.5 ml tubes (Eppendorf, Merk, 306/0421/02) by adding 50 μ l SPA-BAP beads to 25 μ l Buffer 2 and 25 μ l labelled compound (¹²⁵I-BAP 1-40 or test compound). Tubes were then incubated for 180 minutes at room temperature with shaking, followed by centrifugation for 2 minutes at 1000 x g. The supernatants were removed and SPA-BAP pellet washed twice with 300 μ l Buffer 3 containing 1% TWEEN-20 (Sigma, P7949). Non-specific binding for labelled compounds to the SPA beads was determined using incubations as described above but by substituting SPA-BAP beads with SPA-NSB beads. Radioactivity associated with the washed SPA bead pellets was then determined.

The affinity of labelled compounds for fibrillar BAP 1-40 was estimated by subtracting SPA-NSB associated counts from SPA-BAP associated counts. The binding of labelled compounds was then compared to ¹²⁵I-BAP binding, which was

taken as being 100%.

In these experiments, ^{125}I -BAP 1-40 or test compounds were added in equimolar amounts .

Results and discussion

- 5 BAP 1-40 readily self-aggregates. In this assay, the binding of ^{125}I -BAP 1-40 to a fixed amount of amyloid fibrils immobilised on SPA beads was used as a reference for other compounds. Table 1 shows how other amyloid binders and non-binders compare to the binding of ^{125}I -BAP 1-40. The test compounds bind to amyloid fibrils with 40, 15 and 23% the affinity of ^{125}I -BAP 1-40 respectively, 10 which is favourable compared to the ^{125}I -labelled BAP 15-21 sequence (Amersham Biosciences) (21%) and the $^{99\text{m}}\text{Tc}$ -labelled BAP 15-21 sequence (9%). The BAP 15-21 sequence is responsible for the binding of BAP to itself during the formation of amyloid fibrils.

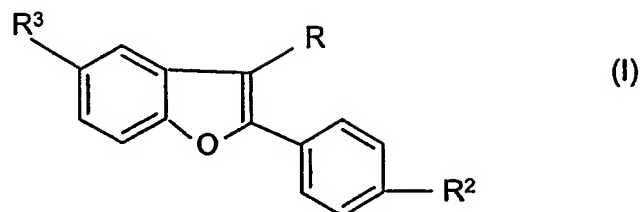
Table 1

15

Compound	Amyloid binding (% of ^{125}I -BAP 1-40)
^{125}I -BAP 1-40	100
Compound of Example 8(iii) Pn216 conjugated	40
Compound of Example 2(vii) Pn216 conjugated	15
Compound of example 1(vii) Pn216 conjugated	23
^{125}I -KKLVFFA (BAP 15-21)	21
$^{99\text{m}}\text{Tc}$ -Pn216-KKLVFF (BAP 15-20)	9

CLAIMS

1. A compound of formula (I):

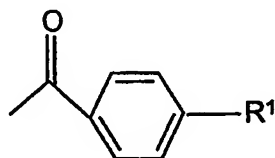


5 or a salt thereof, wherein either:

(iii) R is hydrogen and R² is the group -OCR⁴R⁵[B]-[A] wherein R⁴ and R⁵ are independently selected from hydrogen and C₁-₆ alkyl, [A] is a chelate, and [B] is a linking group and is preferably -C(O)NR⁶- wherein R⁶ is hydrogen or C₁-₆ alkyl;

10 or

(iv) R is the group



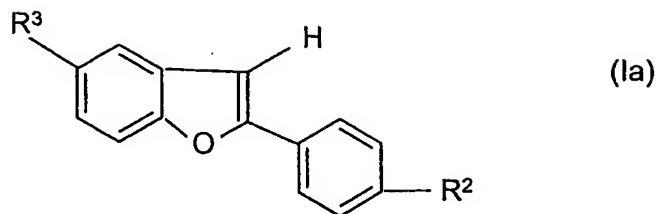
and one of R¹ and R² is the group -OCR⁴R⁵[B]-[A] as defined above; and

the other is the group -(O)ₙ-C₁-₆ alkyl-NR⁷R⁸ wherein n is 0 or 1, and R⁷ and R⁸ are

15 independently selected from hydrogen and C₁-₆ alkyl; and

R³ is halo, preferably chloro.

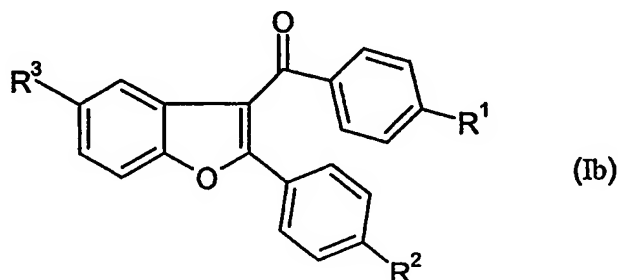
2. A compound according to claim 1 of formula (Ia):



or a salt thereof, wherein R^2 is the group $-\text{OCR}^4\text{R}^5[\text{B}]-[\text{A}]$ wherein R^4 and R^5 are independently selected from hydrogen and C_{1-6} alkyl, $[\text{A}]$ is a chelate, and $[\text{B}]$ is a linking group and is preferably $-\text{C}(\text{O})\text{NR}^6-$ wherein R^6 is hydrogen or C_{1-6} alkyl;

5 and R^3 is halo, preferably chloro.

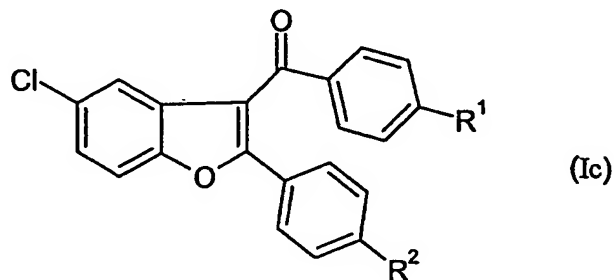
3. A compound according to claim 1 of formula (Ib):



or a salt thereof, wherein:

one of R^1 and R^2 is the group $-\text{OCR}^4\text{R}^5[\text{B}]-[\text{A}]$ as defined in claim 1; and
 10 the other is the group $-(\text{O})_n-\text{C}_{1-6}\text{ alkyl}-\text{NR}^7\text{R}^8$ as defined in claim 1; and
 R^3 is halo, preferably chloro.

4. A compound according to claim 3 of formula (Ic):



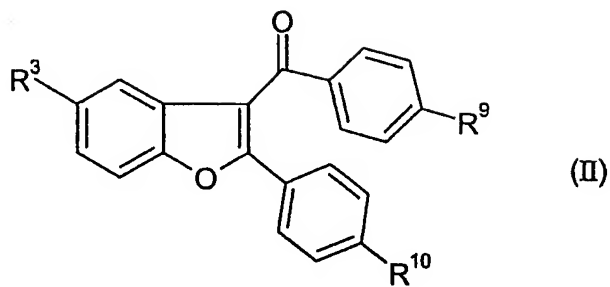
or a salt thereof, wherein:

one of R^1 and R^2 is $-OCH_2C(O)NH-[A]$ wherein $[A]$ is a chelate;

the other is $-O-(CH_2)_{2,3}-NR^7R^8$ wherein R^7 and R^8 are independently selected from hydrogen and C_{1-6} alkyl.

- 5 5. A radiolabelled compound comprising a compound of formula (I), (Ia), (Ib), or (Ic) according to any one of claims 1 to 4, labelled with ^{99m}Tc .
6. A radiopharmaceutical formulation comprising the compound of claim 5 and one or more pharmaceutically acceptable excipients.
7. A kit for the preparation of a radiopharmaceutical comprising a compound of
10 formula (I), (Ia), (Ib), or (Ic) according to any one of claims 1 to 4.
8. A compound according to any one of claims 1 to 5 for use in therapeutic or diagnostic methods.
9. A compound according to any one of claims 1 to 5 for use in *in vivo* imaging or diagnosis of Alzheimer's disease, familial Alzheimer's disease, type II diabetes,
15 Down's syndrome, homozygotes for the apolipoprotein E4 allele, rheumatoid arthritis, systemic amyloidosis (primary and secondary), and haemorrhagic stroke.
10. Use of a compound according to any one of claims 1 to 5 in the manufacture of a medicament for *in vivo* imaging or diagnosis of Alzheimer's disease, familial Alzheimer's disease, type II diabetes, Down's syndrome, homozygotes for the
20 apolipoprotein E4 allele, rheumatoid arthritis, systemic amyloidosis (primary and secondary), and haemorrhagic stroke.
11. A method for *in vivo* imaging or diagnosis of Alzheimer's disease, familial Alzheimer's disease, type II diabetes, Down's syndrome, homozygotes for the apolipoprotein E4 allele, rheumatoid arthritis, systemic amyloidosis (primary and
25 secondary), and haemorrhagic stroke which comprises administration of a compound of claim 5.

12. A compound of formula (II)



or a protected derivative thereof, wherein one of R^9 and R^{10} is $-\text{OCR}^4\text{R}^5\text{C}(\text{O})\text{OH}$ wherein R^4 and R^5 are as defined in claim 1; and

- 5 the other is $-(\text{O})_n\text{-C}_{1-6}\text{alkyl-NR}^7\text{R}^8$ wherein n , R^7 , and R^8 are as defined in claim 1; and R^3 is as defined in claim 1.

INTERNATIONAL SEARCH REPORT

In tional Application No

PCT/GB 02/05641

A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C07D307/78 A61K51/04

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C07D A61K

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, PAJ, CHEM ABS Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	WO 95 17382 A (LILLY CO ELI) 29 June 1995 (1995-06-29) claims ---	1-12
A	US 3 947 470 A (BRENNER L MARTIN ET AL) 30 March 1976 (1976-03-30) cited in the application examples ---	1-12
A	US 4 024 273 A (BRENNER L MARTIN ET AL) 17 May 1977 (1977-05-17) cited in the application examples ---	1-12
-/--		



Further documents are listed in the continuation of box C.



Patent family members are listed in annex.

* Special categories of cited documents:

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- *E* earlier document but published on or after the international filing date
- *L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- *O* document referring to an oral disclosure, use, exhibition or other means
- *P* document published prior to the international filing date but later than the priority date claimed

- *T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- *X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- *Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- *&* document member of the same patent family

Date of the actual completion of the international search

25 February 2003

Date of mailing of the international search report

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Name and mailing address of the ISA

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INTERNATIONAL SEARCH REPORT

Int tional Application No

PCT/GB 02/05641

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	<p>ALLSOP D ET AL: "3-p-toluoyl-2-'4'-(3-diethylaminopropoxy)-phenyl-benzofuran and 2-'4'-(3-diethylaminopropoxy)-phenyl-benzofuran do not act as surfactants or micelles when inhibiting the aggregation of beta-amyloid peptide" BIOORGANIC & MEDICINAL CHEMISTRY LETTERS, OXFORD, GB, vol. 11, no. 2, January 2001 (2001-01), pages 255-257, XP004314861 ISSN: 0960-894X the whole document</p> <p>---</p>	1-12
A	<p>TWYMAN L J ET AL: "A Short Synthesis of the beta-amyloid (Abeta) Aggregation Inhibitor 3-p-Toluoyl-2-'4'-(3-diethylaminopropoxy)-phenyl-benzofuran" TETRAHEDRON LETTERS, ELSEVIER SCIENCE PUBLISHERS, AMSTERDAM, NL, vol. 40, no. 52, 24 December 1999 (1999-12-24), pages 9383-9384, XP004183675 ISSN: 0040-4039 the whole document</p> <p>---</p>	1-12
A	<p>HOWLETT, DAVID R. ET AL: "Inhibition of fibril formation in beta-amyloid peptide by a novel serie of benzofurans" BIOCHEMICAL JOURNAL (1999), 340(1), 283-289 , XP009006358 cited in the application page 285</p> <p>-----</p>	1-12

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Information on patent family members

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